

Set # 3: Name the following mixed compounds.

- 1.) PbSO_4 ^{Lead (II)} sulfate ionic
- 2.) AlF_3 ^{aluminum} fluoride ionic
- 3.) BaO_2 ^{barium} oxide ionic
- 4.) N_2O_5 ^{dinitrogen} pentoxide covalent
- 5.) HClO ^{hypochlorous} acid acid
- 6.) MgBr_2 ^{magnesium} bromide ionic
- 7.) $\text{Ag}_2\text{Cr}_2\text{O}_7$ ^{silver} dichromate ionic
- 8.) ZnCl_2 ^{zinc} chloride ionic
- 9.) $\text{Fe}(\text{OH})_3$ ^{Iron (III)} hydroxide ionic
- 10.) H_3PO_4 ^{phosphoric} acid acid