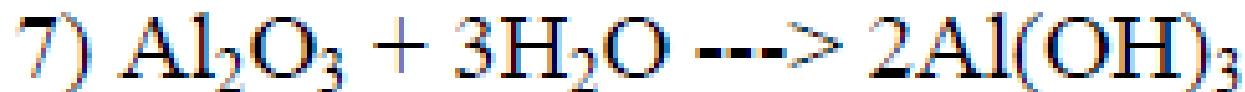
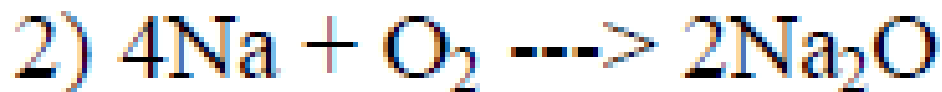
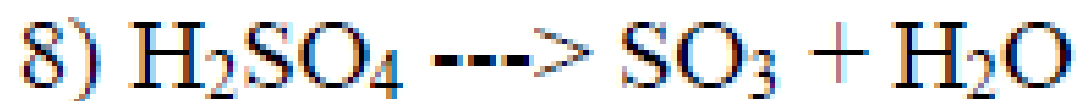
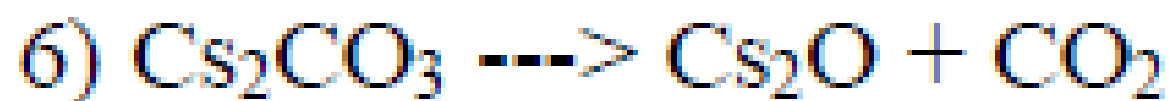
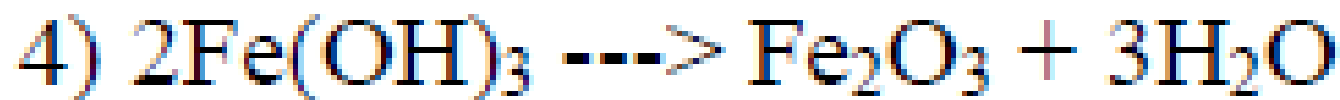
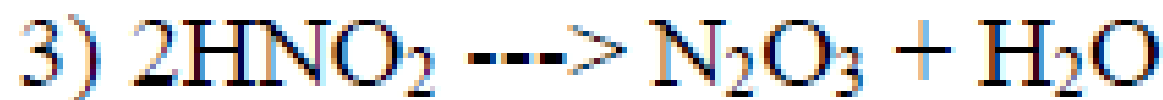


## Combination- Answers



## Decomposition- Answers



## Single Replacement- Answers



If Fe(III) is used, then  $\text{Fe}(\text{C}_2\text{H}_3\text{O}_2)_3$  would result.



If Fe(III) is used, then  $\text{Fe}_2(\text{SO}_4)_3$  would result.



## Combustion- Answers

